Biosynthesis of α -Cyclopiazonic Acid: Stereochemical Aspects of D-ring Formation

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(3*R*)-Mevalonic acid lactone is incorporated into α -cyclopiazonic acid with retention of the 4-*pro-R*-hydrogen atom; [1,2-¹³C₂]acetate has been used as a probe to show that the cyclization of β -cyclopiazonic acid into α -cyclopiazonic acid involves a *syn* addition to the double bond.

The biosynthesis of α -cyclopiazonic acid (1) has received considerable attention.¹⁻⁶ Mevalonic acid, acetic acid, and L-tryptophan have been implicated as the biogenetic precursors,^{1,2} and the biogenesis has been shown to occur *via*

cyclization of β -cyclopiazonic acid (2).¹⁻⁵ A ping-pong bi-bi mechanism has been suggested for this cyclization⁴ which occurs with loss of the 3-*pro-S*-hydrogen atom from tryptophan.⁵ In this paper we report studies on the stereochemistry

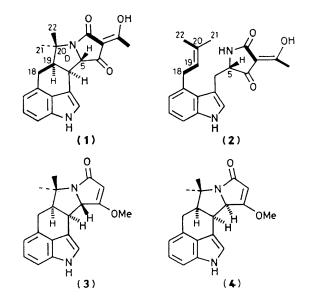


Table 1. Incorporation of labelled mevalonic acid lactones into α -cyclopiazonic acid (1).

	³ H: ¹⁴ C ratio	³ H: ¹⁴ C ratio in (1)
$ \begin{array}{c} (3S,4S) - [4^{-3}H] + (3R,4R) - [4^{-3}H] + \\ (3RS) - [2^{-14}C] \\ (3S,4S) - [4^{-3}H] + (3R,4R) - [4^{-3}H] + \\ (3R) - [3^{-14}C] \end{array} \right\} $	2.35	2.44
$(3S,4S)-[4-^{3}H] + (3R,4R)-[4-^{3}H] + $ $(3R)-[3-^{14}C]$	1.67	0.83

of the cyclization of β -cyclopiazonic acid to α -cyclopiazonic acid with regard to the incorporation of the mevalonic acid unit and the disposition of the geminal dimethyl groups in α - and β -cyclopiazonic acids.

Samples of stereospecifically labelled $[4-{}^{3}H]$ mevalonic acid lactones, mixed with $(3RS)-[2-{}^{14}C]-$ and $(3R)-[3-{}^{14}C]$ -mevalonic acid lactone (Table 1), were fed to cultures of *Penicillium* griseofulvum Dierckx, C.S.I.R. 1082. Analysis of the label found in the purified α -cyclopiazonic acid indicated that the (3R)-isomer of mevalonic acid lactone is incorporated into (1) with retention of the 4-pro-R-hydrogen atom.

Investigation of the stereochemical origin of the geminal methyl groups [C(21)] and C(22) in (1) is hampered by their equivalence to chemical reaction and the similarity of their chemical shifts in ¹H and ¹³C n.m.r. spectroscopy, thus making unambiguous assignments impossible (Table 2). However, treatment of (1) with 0.05M sulphuric acid in methanol, followed by methylation with diazomethane, yields a mixture of O-methyldesacetylcyclopiazonic acid (3) and its 5-epimer (4).¹ In the 13 C n.m.r. spectrum of the latter compound the two methyl groups C(21) and C(22) are assignable (Table 2). In the conformation of the normal series both methyl groups experience three γ -gauche interactions whereas in the 5-epimer conformation the β -methyl group experiences four γ -gauche and the α -methyl group two γ -gauche interactions. The resonance of the β -methyl group will, therefore, shift to higher field with respect to its normal series conformation resonance, whereas the resonance of the α -methyl group will shift to lower field.

The assignment of the methyl resonances [C(21) and C(22)] in the ¹³C n.m.r. spectrum of (2) is straightforward as a large body of opinion places the chemical shift of the (Z)-methyl

Table 2.¹³C N.m.r. data^a for compounds (1)--(4).

		(1)		(2)		(3)		(4)
Carbon	່ຈ	¹ <i>J</i> (C,C)	b ິδ	$^{1}J(C,C)^{\mu}$	δ	${}^{1}J(C,C)^{1}$	δ	$^{1}J(C,C)^{b}$
18	26.6	34.8	32.4	43.7	26.8	34.5	26.5	36.0
19	53.1	34.7	123.5	43.8	54.4	34.5	53.9	36.1
20	63.5	39.8	132.9	42.5	61.6	40.4	61.2	36.6
21	26.4		25.7		25.7		30.8	
22	24.5	40.0	18.1	41.7	25.2	40.5	20.9	36.6

^a δ in p.p.m. relative to internal Me₄Si in CDCl₃, J in Hz. ^b Obtained for compounds derived from culture media supplemented with [1,2-¹³C₂]acetate.

group at substantially higher field than that of the (E)-methyl group for dimethylallyl moieties in numerous model systems.⁷ Moreover, the intact double labels observed here confirm this assignment if the side-chain results from a normal isoprene pathway.⁸

Incorporation of $[1,2^{-13}C_2]$ acetate into both (1) and (2), followed by chemical modification of (1) into (4), and analysis of the ¹³C n.m.r. spectra of (2) and (4), allow the intact acetate units in these compounds to be identified (Table 2). As the stereochemistry of the geminal dimethyl groups is not affected in the transformation (1) \rightarrow (4), the intact acetate units in (1) and (2) are as indicated in heavy print on the structural formulae. Thus $[1,2^{-13}C_2]$ acetate may be used as an effective probe for the stereochemistry of the geminal dimethyl groups in the conversion of (2) into (1). As can be seen, the (Z)methyl group C(22) in β -cyclopiazonic acid (2) becomes the β -methyl group C(22) in α -cyclopiazonic acid (1). Taken together with the established relative configuration at C(19) in (1),⁹ this corresponds to a *syn* addition to the double bond in the transformation of (2) into (1).

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